# **FIITJEE** INTERNAL Common Test

# **PHYSICS, CHEMISTRY & MATHEMATICS**

## QP CODE: 100856

Common Test-5

Time Allotted: 3 Hours

Maximum Marks: 180

- Please read the instructions carefully. You are allotted 5 minutes specifically for this purpose.
- You are not allowed to leave the Examination Hall before the end of the test.

### INSTRUCTIONS

Caution: Question Paper CODE as given above MUST be correctly marked in the answer OMR sheet before attempting the paper. Wrong CODE or no CODE will give wrong results.

#### **A. General Instructions**

- 1. Attempt ALL the questions. Answers have to be marked on the OMR sheets.
- 2. This question paper contains Three Sections.
- 3. Section-I is Physics, Section-II is Chemistry and Section-III is Mathematics.
- 4. All the section can be filled in PART-A & B of OMR.
- 5. Rough spaces are provided for rough work inside the question paper. No additional sheets will be provided for rough work.
- 6. Blank Papers, clip boards, log tables, slide rule, calculator, cellular phones, pagers and electronic devices, in any form, are not allowed.

#### B. Filling of OMR Sheet

- 1. Ensure matching of OMR sheet with the Question paper before you start marking your answers on OMR sheet.
- 2. On the OMR sheet, darken the appropriate bubble with **Blue/Black Ball Point Pen** for each character of your Enrolment No. and write in ink your Name, Test Centre and other details at the designated places.
- 3. OMR sheet contains alphabets, numerals & special characters for marking answers.

#### C. Marking Scheme For All Two Parts.

- (i) **Part-A (01-04)** Contains Four (04) multiple choice questions which have ONLY ONE CORRECT answer Each question carries +3 marks for correct answer and -1 marks for wrong answer.
- (ii) PART-A (05-07) contains (3) Multiple Choice Questions which have One or More Than One Correct answer.

*Full Marks*: +4 If only the bubble(s) corresponding to all the correct options(s) is (are) darkened. *Partial Marks*: +1 For darkening a bubble corresponding to each correct option, provided NO incorrect option is darkened.

Zero Marks: 0 If none of the bubbles is darkened.

Negative Marks: -1 In all other cases.

For example, if (A), (C) and (D) are all the correct options for a question, darkening all these three will result in +4 marks; darkening only (A) and (D) will result in +2 marks; and darkening (A) and (B) will result in -1 marks, as a wrong option is also darkened.

(iii) Part-A (08-11) – This section contains Four (04) Matching List Sets. Each set has ONE Multiple Choice Question. Each set has TWO lists: List-I and List-II. List-I has Four entries (P), (Q), (R) and (S) and List-II has Five entries (1), (2), (3), (4) and (5). FOUR options are given in each Multiple Choice Question based on List-I and List-II and ONLY ONE of these four options satisfies the condition asked in the Multiple Choice Question. Each question carries +3 Marks for correct answer and -1 marks for wrong answer.

(iii) Part-B (01-06) This section contains SIX (06) questions. The answer to each question is a NON-NEGATIVE INTEGER. For each question, enter the correct integer corresponding to the answer. Each question carries +4 marks for correct answer. There is no negative marking.

Name of the Candidate:	
Batch:	Date of Examination:
Enrolment Number:	

# <u>SECTION - I: PHYSICS</u>

### (PART – A)

## (Single Correct Answer Type)

This section contains **4 multiple choice questions**. Each question has four choices (A), (B), (C) and (D) out of which **ONLY ONE is correct**.

1. A geostationary satellite is orbiting the earth at a height of 6 R above the surface of earth, R being the radius of earth. The time period of another satellite at a height of 2.5 R from the surface of earth is

(A)  $6/\sqrt{2}$  hr (B) 10 Hr (C) 6 hr (D)  $6\sqrt{2}$  hr

 A body is projected vertically upwards from the earth's surface to reach a height 7R, where R is the Radius and M is mass of earth. The minimum velocity required to do so is

(A) 7GM	(P) 7GM	(C) 8GM	(D) 20 GM
$(A) \sqrt{8R}$	$(D) \sqrt{4R}$	(C) <u>√</u> 3R	(D) <u>11</u> R

3. A spherical hollow is made in a lead sphere of radius R such that its surface touches the outside surface of the lead sphere and passes through it centre. Mass of lead sphere before hollowing is M. A particle of mass 'm' is placed at a distance 'd' from the centre of the lead sphere and on the lien joining the centre of the sphere and the centre of the hollow as shown. Force of attraction between the hallowed sphere and the particle



Space For Rough Work

A simple pendulum has a time period T<sub>1</sub> when on earth's surface and T<sub>2</sub> when taken to a 4. height R above the earth's surface. R is the radius of the earth. The value of  $T_2/T_1$  is (B) √2 (A) 1 (C) 4 (D) 2

(One or More Than One Options Correct Type) This section contains 3 multiple choice questions. Each question has 4 choices (A), (B), (C) and (D), out of which ONE or MORE THAN ONE is correct.

5. Inside a uniform spherical shell: (A) potential is zero (B) field is zero (C) potential is constant (D) field is constant

The magnitude of gravitational field at distance r<sub>1</sub> and r<sub>2</sub> from the centre of a uniform sphere 6. of radius R and mass M are I<sub>1</sub> and I<sub>2</sub> respectively. Then

(B)  $\frac{l_1}{l_2} = \frac{r_2^2}{r_1^2}$  if  $r_1 > R$  and  $r_2 > R$ (A)  $\frac{l_1}{l_2} = \frac{r_1}{r_2}$  if  $r_1 < R$  and  $r_2 < R$ (C)  $\frac{l_1}{l_2} = \frac{r_1}{r_2}$  if  $r_1 > R$  and  $r_2 > R$ (D)  $\frac{l_1}{l_2} = \frac{r_1^2}{r_2^2}$  if  $r_1 < R$  and  $r_2 < R$ 

7. A planet is moving round the sun in an elliptical orbit as shown. As the planet moves from A to B (A) its kinetic energy will decrease

- (B) its potential energy will remain unchanged
- (C) its angular momentum about centre of sun will remain unchanged
- (D) its speed is minimum at A

Space For Rough Work

В S

#### (Matching List Sets)

This section contains FOUR (04) Matching List Sets. Each set has ONE Multiple Choice Question. Each set has TWO lists: List-I and List-II. List-I has Four entries (P), (Q), (R) and (S) and List-II has Five entries (1), (2), (3), (4) and (5). FOUR options are given in each Multiple Choice Question based on List-I and List-II and **ONLY ONE** of these four options satisfies the condition asked in the Multiple Choice Question.

8. A planet of mass M, has two natural satellites with masses  $m_1$  and  $m_2$ . The radii of their circular orbits are R1 and R2 respectively. Ignore the gravitational force between the satellites. Define  $v_1$ ,  $L_1$ ,  $K_1$  and  $T_1$  to be, respectively, the orbital speed, angular momentum, kinetic energy and the time period of revolution of satellite 1; and v<sub>2</sub>, L<sub>2</sub>, K<sub>2</sub> and T<sub>2</sub> to be the corresponding quantities of satellite 2. Given  $m_1/m_2 = 2$  and  $R_1/R_2 = 1/4$ , match the ratios in List-I to the numbers in List-II.

List-I			List- <mark>II</mark>				
(P)	V <sub>1</sub>	(1)	1				
	$\overline{V_2}$		8				
(Q)	L	(2)	1				
	$\overline{L_2}$						
(R)	K,	(3)	2				
	$\overline{K_2}$						
(S)	Т	(4)	8				
	$\overline{T_2}$						

The correct option is:

 $\begin{array}{l} (A) \ P \rightarrow 4 \ ; \ Q \rightarrow 2 \ ; \ R \rightarrow 1 \ ; \ S \rightarrow 3 \\ (C) \ P \rightarrow 2 \ ; \ Q \rightarrow 3 \ ; \ R \rightarrow 1 \ ; \ S \rightarrow 4 \end{array} \qquad \begin{array}{l} (B) \ P \rightarrow 3 \ ; \ Q \rightarrow 2 \ ; \ R \rightarrow 4 \ ; \ S \rightarrow 1 \\ (D) \ P \rightarrow 2 \ ; \ Q \rightarrow 3 \ ; \ R \rightarrow 4 \ ; \ S \rightarrow 1 \end{array}$ 

9. For two particles describing circular path under mutual gravitational force of attraction.

List-I			List-II		
(P)	If speed of the particles decreases such that velocity of centre of mass remains zero, then	(1)	particle separation will increase		
(Q)	If one of the particles stops then	(2)	particle separation will decrease		
(R)	If both particle stops for a moment then	(3)	common centre will lie on centre of mass		
(S)	If speed of both particles increase such that velocity of centre of mass remains zero.	(4)	particle collide at centre of mass		

The correct option is:

(A)  $P \rightarrow 1,4$ ;  $Q \rightarrow 1$ ;  $R \rightarrow 2,3$ ;  $S \rightarrow 3$ (B)  $P \rightarrow 1,2$ ;  $Q \rightarrow 3$ ;  $R \rightarrow 2,4$ ;  $S \rightarrow 1,3$ (C)  $P \rightarrow 1,2$ ;  $Q \rightarrow 4$ ;  $R \rightarrow 2,3$ ;  $S \rightarrow 1,4$  (D)  $P \rightarrow 1,3$ ;  $Q \rightarrow 1$ ;  $R \rightarrow 2,4$ ;  $S \rightarrow 2,3$ 

10. Gravitational potential on the surface of an isolated uniform solid sphere of mass M and radius R is found to be V<sub>0</sub>. A spherical cavity having radius R/2 is created inside the sphere which is touching the surface of original sphere. The cavity is then filled with material having density 16 times that of original sphere. A, B, C and D are consecutive points as shown in the figure each R/2 apart. V<sub>A</sub>, V<sub>B</sub>, V<sub>C</sub> and V<sub>D</sub> are gravitational potentials found at points A, B, C and D respectively.



	List-I		List-I	I	
(P)	VA	(1)	7V <sub>0</sub>		
(Q)	VB	(2)	13/4 V <sub>0</sub>		
(R)	Vc	(3)	19/4 V <sub>0</sub>		
(S)	V <sub>D</sub>	(4)	21/4 V <sub>0</sub>		
		(5)	25/ <mark>4 V</mark> 0		

The correct option is:

(A)  $P \rightarrow 3$ ;  $Q \rightarrow 1$ ;  $R \rightarrow 4$ ;  $S \rightarrow 2$ (C)  $P \rightarrow 3$ ;  $Q \rightarrow 1$ ;  $R \rightarrow 2$ ;  $S \rightarrow 4$  (B)  $P \rightarrow 3$ ;  $Q \rightarrow 5$ ;  $R \rightarrow 2$ ;  $S \rightarrow 1$ (D)  $P \rightarrow 2$ ;  $Q \rightarrow 3$ ;  $R \rightarrow 4$ ;  $S \rightarrow 5$ 

11. Consider a planet of radius r having density  $\lambda$ . A tunnel is dug inside it at a distance r/2 from its centre as shown. An object of mass m is left in the tunnel at the surface at t = 0 then



	List-I		List-II
(P)	Time taken by the object to reach the mid of the	(1)	Zero
	tunnel		
(Q)	Magnitude of velocity of object at the centre of	(2)	g
	the tunnel	、 <i>,</i>	
(R)	Normal reaction applied by wall of the tunnel on	(3)	$\sqrt{3\pi}$
	the object	. ,	$\sqrt{16G\lambda}$
(S)	Acceleration of object when it reach the mid of	(4)	2 _ () = =
. /	the tunnel	. ,	-πολητη 3
		(5)	$\left(\sqrt{\pi 6\lambda}\right)$ r

The correct option is:

(A)  $P \rightarrow 3$ ;  $Q \rightarrow 1$ ;  $R \rightarrow 5$ ;  $S \rightarrow 2$ (C)  $P \rightarrow 4$ ;  $Q \rightarrow 5$ ;  $R \rightarrow 2$ ;  $S \rightarrow 3$   $\begin{array}{l} (B) \ P \rightarrow 3 \ ; \ Q \rightarrow 5 \ ; \ R \rightarrow 4 \ ; \ S \rightarrow 1 \\ (D) \ P \rightarrow 3 \ ; \ Q \rightarrow 2 \ ; \ R \rightarrow 1 \ ; \ S \rightarrow 4 \end{array}$ 

### (PART – B)

#### (Non – Negative Integer)

- 1. A satellite is moving in a circular orbit around earth at a height R above earth surface (R being radius of earth). It's velocity should be increased to K times its initial orbital speed value, so as to make it escape from earth gravitational pull and reach infinity. Find the value of "K<sup>2</sup>"
- 2. If the mass of the planet that has a satellite whose time period is T and orbital radius r is  $\frac{K\pi^2 r^3}{4GT^2}$ , then the value of 'K' is
- 3. A satellite is revolving round the earth in a circular orbit of radius "r" and velocity V<sub>o</sub>. A body is projected from the satellite in forward direction with relative velocity  $V_{rel} = \left(\sqrt{\frac{5}{4}} 1\right)V_{o}$ .

If ratio of minimum and maximum distances from earth's centre during subsequent motion of the particle is K, then value of 10 K will be?

4. Three particles, each of mass m, are situated at the vertices of an equilateral triangle of side length a. The only forces acting on the particles are their mutual gravitational forces. It is desired that each particle moves in a circle while maintaining the original mutual separation

a. Find the initial velocity that should be given to each particle. (take a =  $\frac{GM}{16}$ )

- 5. A rocket is fired form the earth to the moon. The distance between the earth and the moon is r and the mass of the earth is 81 times the mass of the moon. The gravitational force on the rocket will be zero, when its distance from the moon is  $\mu$ r. Find the value of '10 $\mu$ '.
- 6. A uniform ring of mass m and radius r is placed directly above a uniform sphere of mass M and of equal radius. The centre of the ring is directly above the centre of the sphere at a distance  $r\sqrt{3}$  as shown in the figure. The gravitational force exerted by the sphere on the ring will be  $\sqrt{k} \frac{GMm}{8r^2}$  then



find the value of 'k'.

## <u>SECTION – II: CHEMISTRY</u>

(PART – A)

#### (Single Correct Answer Type)

This section contains **4 multiple choice questions**. Each question has four choices (A), (B), (C) and (D) out of which **ONLY ONE is correct**.

1. In a chemical reaction,  $A(g) + 2B(g) \xrightarrow{K_C} 2C(g) + D(g)$ , the initial concentration of B was 1.5 times of the concentration of A, but the equilibrium concentrations of A and B were found to be equal. The equilibrium constant(K<sub>c</sub>) for the aforesaid chemical reaction is (A) 16 (B) 4 (C) 1 (D) 1/4

(B)  $K_p = \frac{K_c}{(a+b)^2}$ 

(D)  $K_{p} = K_{c}$ 

2. The correct relations between K<sub>P</sub> and K<sub>c</sub> for the reaction  $aX + bY \Longrightarrow bZ + aW$  is

(A)  $K_p = K_c[RT]^{a+b}$ 

(C)  $K_p = K_c[RT]$ 

3. The ratio of degree of dissociation of  $HCN(K_a = 10^{-9})$  in its 0.1 M and 0.001 M solution: (A) 0.1 (B) 0.2 (C) 0.3 (D) 0.4

4.  $pK_a$  of a weak acid(HA) and  $pK_b$  of a weak base(BOH) are 3.2 and 3.4 respectively. The pH of their salt(AB) solution is

(B) 6.9

(D) 1.0

- (A) 7.2
- (C) 7.0

#### (One or More Than One Options Correct Type)

This section contains 3 multiple choice questions. Each question has 4 choices (A), (B), (C) and (D), out of which ONE or MORE THAN ONE is correct.

5. The equilibrium  $SO_2Cl_2(g) \Longrightarrow SO_2(g) + Cl_2(g)$  is attained at 25°C in a closed container at constant P. An inert gas, helium is introduced now. Which of the following statements are incorrect when equilibrium is restablished? (A) Concentrations of SO<sub>2</sub>, Cl<sub>2</sub> and SO<sub>2</sub>Cl<sub>2</sub> change

A) Concentrations of  $SO_2$ ,  $CI_2$  and  $SO_2C$ 

- (B) More Cl<sub>2</sub> is formed
- (C) Concentration of SO<sub>2</sub> is reduced
- (D) More SO<sub>2</sub>Cl<sub>2</sub> is formed
- 6. Which are buffer mixtures?
  (A) HCN and NaCN
  (C) CH<sub>3</sub>COONa and CH<sub>3</sub>COOH

(B) NaOH and NaNO<sub>3</sub> (D) NH<sub>4</sub>OH and NH<sub>4</sub>Cl

7. The given aqueous solution at 25°C is

(A) acidic if  $[H^+] < \sqrt{K_w}$ (C) acidic if  $[H^+] > \sqrt{K_w}$  (B) alkaline if  $[H^+] < \sqrt{K_w}$ (D) neutral if  $[H^+] = \sqrt{K_w}$ 

#### (Matching List Sets)

This section contains **FOUR (04)** Matching List Sets. Each set has **ONE** Multiple Choice Question. Each set has **TWO** lists: List-I and List-II. List-I has Four entries (P), (Q), (R) and (S) and List-II has Five entries (1), (2), (3), (4) and (5). FOUR options are given in each Multiple Choice Question based on List-I and List-II and ONLY ONE of these four options satisfies the condition asked in the Multiple Choice Question.

8. Match the lists.

List – I			List– II			
(P)	$CaCO_3(s) \Longrightarrow CaO(s) + CO_2(g)$	(1)	$K_p > K_c$ above room temperature			
(Q)	$CO(g) + Cl_2(g) \Longrightarrow COCl_2(g)$	(2)	K <sub>p</sub> = K <sub>c</sub> above room temperature			
(R)	$H_2(g) + I_2(g) \Longrightarrow 2HI(g)$	(3)	K <sub>p</sub> < K <sub>c</sub> above room temperature			
(S)	$HCI(g) \Longrightarrow H^{+}(aq) + CI^{-}(aq)$	(4)	$K_{p}$ and $K_{c}$ not defined			
		(5)	$K_{p} = \frac{K_{c}}{C}$ change with temperature			
(A) P	$\rightarrow$ 1; Q $\rightarrow$ 3; R $\rightarrow$ 2; S $\rightarrow$ 4	(B) P	$\rightarrow$ 3; Q $\rightarrow$ 4; R $\rightarrow$ 2; S $\rightarrow$ 1			
(C) P	$\rightarrow$ 4; Q $\rightarrow$ 2; R $\rightarrow$ 1; S $\rightarrow$ 3	(D) P	$\rightarrow$ 2; Q $\rightarrow$ 3; R $\rightarrow$ 1; S $\rightarrow$ 4			

9. Match the lists.

	List – I	List– II		
(P)	Pressure increased in	(1)	Equilibrium shifted in forward	
	$2NO(g) \Longrightarrow N_2(g) + O_2(g)$		direction	
(Q)	Pressure increased in	(2)	Equilibrium shifted in backward	
	$CH_4(g) + H_2O(\ell) = CO(g) + 3H_2(g)$		direction	
(R)	Temperature increased and pressure	(3)	Equilibrium remains unaffected	
	increased			
	3O <sub>2</sub> (g) = 2O <sub>3</sub> (g); ∆H = 285kJ			
(S)	Pressure decreased and mole of N <sub>2</sub> increased	(4)	Theoretically we cannot predict	
	$N_2(g) + 2O_2(g) \Longrightarrow 2NO_2(g); \Delta H = 66.4 kJ$			
		(5)	Equilibrium initially shift in	
			backward direction then shift in forward direction	
	$(\mathbf{P}) \mathbf{P} \rightarrow 2$			
(A) P	$\rightarrow$ 5; Q $\rightarrow$ 3; R $\rightarrow$ 2; S $\rightarrow$ 4 (B) P $\rightarrow$ 3	; Q →	$2; R \rightarrow 1; S \rightarrow 4$	
(C) P	$\rightarrow$ 4; Q $\rightarrow$ 5; R $\rightarrow$ 1; S $\rightarrow$ 3 (D) P $\rightarrow$ 1	; $Q \rightarrow$	$\sim 2; R \rightarrow 4; S \rightarrow 3$	

10. Match the lists.

List – I (Salts)			List– II (Solubility products)			
(P)	Mercurous iodide	(1)	108 S⁵			
(Q)	Aluminium phosphate	(2)	4 S <sup>3</sup>			
(R)	Calcium phosphate	(3)	S <sup>2</sup>			
(S)	Zirconium(IV) phosphate	(4)	6912 S <sup>7</sup>			
		(5)	27 S <sup>4</sup>			
(A) P (C) P	$\rightarrow 2; Q \rightarrow 3; R \rightarrow 1; S \rightarrow 4$ $\rightarrow 2; Q \rightarrow 1; R \rightarrow 4; S \rightarrow 5$	(B) P (D) P	$\rightarrow$ 4; Q $\rightarrow$ 5; R $\rightarrow$ 1; S $\rightarrow$ 4; Q $\rightarrow$ 2; R $\rightarrow$ 5; S $\rightarrow$	$\rightarrow 2$ → 3		

#### 11. Match the lists.

matori						
List – I (Salts)			List– II (pH of aqueous solution)			
(P)	Salt of weak acid and weak base	(1)	$pH = \frac{1}{2} \left[ pK_w + pK_a + \log C \right]$			
(Q)	Salt of weak acid and strong base	(2)	pH <mark>= 1/2[pK<sub>w</sub> + pK<sub>a</sub> – pK<sub>b</sub>]</mark>			
(R)	Salt of strong acid and strong base	(3)	$pH = \frac{1}{2} \left[ pK_w - pK_b - \log C \right]$			
(S)	Salt of strong acid and weak base	(4)	$pH = 1/2[pK_w]$			
		(5)	$pH = 1/2[pK_a - logC]$			
(A) P	$\rightarrow$ 1; Q $\rightarrow$ 2; R $\rightarrow$ 3; S $\rightarrow$ 4	(B) P	$\rightarrow$ 2; Q $\rightarrow$ 5; R $\rightarrow$ 1; S $\rightarrow$ 4			
(C) P	$\rightarrow$ 1; Q $\rightarrow$ 5; R $\rightarrow$ 2; S $\rightarrow$ 3	(D) P	$\rightarrow$ 2; Q → 1; R → 4; S → 3			

(PART – B)

(Non – Negative Integer)

- 1. In the reaction:  $C(s) + CO_2(g) = 2CO(g)$ , the equilibrium pressure is 6 atm. If 50% of CO<sub>2</sub> reacts than K<sub>P</sub> of the reaction is....
- 2. For the reaction A(g) = B(g) at 495 K,  $\Delta_r G^\circ = -9.478$  kJ mol<sup>-1</sup>. If we start the reaction in a closed container at 495 K with 22 millimoles of A, the amount of B in the equilibrium mixture is \_\_\_\_\_ millimoles(Round off to the nearest integer). [R = 8.314 J mol<sup>-1</sup> K<sup>-1</sup>; In 10 = 2.303]

- 3. The pH of a solution obtained by mixing 10 mL of 0.45 M HCl and 40 mL of 0.1 M NaOH is\_\_\_\_\_
- A certain buffer solution equals concentration of X<sup>-</sup> and HX. K<sub>b</sub> for X<sup>-</sup> is 10<sup>-10</sup>. The pH of buffer is \_\_\_\_\_
- 5. What is the minimum pH required for the precipitation of ZnS in a solution from 0.01 M ZnCl<sub>2</sub> saturated with 0.10 M H<sub>2</sub>S? (Given  $K_{sp(ZnS)} = 10^{-21} K_{a_1} \times K_{a_2} = 10^{-20}$  for H<sub>2</sub>S)
- 6. A mixture of A(g), B(g) and C(g) at equilibrium has a average molecular weight 80. According to the reaction

 $A(g) \Longrightarrow B(g) + C(g)$ 

The degree of dissociation of A(g) is  $\_\_\_ \times 10^{-2}$ : (Given: mol. wt of A = 100; mol.wt. of B = 60; mol wt of C - 40)

# **SECTION - III: MATHEMATICS**

(PART – A)						
This se of whic	(Single Correct ection contains 4 multiple choice questions. Ea h ONLY ONE is correct.	Answer Type) ach question has four choices (A), (B), (C) and (D) out				
1. The sequence loga, $\log \frac{a^2}{b}$ , $\log \frac{a^3}{b^2}$ , is						
	(A) a G.P (C) an H.P	(B) an A.P (D) none of these				
2.	The number of terms common to the two A. 3+8+13+18+23+198	Ps 2+5+8+11++98 and				
	(A) 33 (C) 7	(B) 40 (D) none of these				
3.	If the fifth term of a G.P. is 2, then the produ (A) 256 (C) 1024	uct of its firs <mark>t 9 terms is</mark> (B) 512 (D) None of these				
4.	The eccentricity of the ellipse $\frac{x^2}{a^2+2} + \frac{y^2}{a^2+3}$	$\frac{1}{3} = 1$ is $\frac{1}{3}$ then Length of latus rectum is				
	(A) $\frac{3}{\sqrt{2}}$	(B) $\frac{3}{2}$				
	(C) $\frac{16}{3}$	(D) none of these				
This s and (D)	One or More Than One) ection contains 3 multiple choice questio ), out of whi <mark>ch ONE or MORE THAN ONE is cor</mark> t	<b>Options Correct Type)</b> ns. Each question has 4 choices (A), (B), (C) rect.				
5.	The equation of the conic with focus at eccentricity $\sqrt{2}$ is	(1, -1), directrix along x $-y + 1 = 0$ and with				
	(A) $x^2 - y^2 = 1$	(B) $xy = 1$ (D) $2xy + 4x + 4y + 1 = 0$				
	(0) 2xy - 4x + 4y + 1 = 0	(D) 2xy + 4x - 4y - 1 = 0				

6. Equation of tangent to the ellipse  $\frac{x^2}{25} + \frac{y^2}{16} = 1$  which cuts off equal intercepts on the axes is

(A) 
$$x + y - \sqrt{41} = 0$$
 (B)  $x - y + \sqrt{41} = 0$  (C)  $x + y - 9 = 0$  (D)  $x - y + 9 = 0$ 

 $\mathbf{a}$ 

7. If 
$$(x+2)^2 + (y-4)^2 = k^2 \frac{(3x+y+2)^2}{100}$$
 represents a hyperbola then the integral value of k can be  
(A) 2 (B) 3 (C) 4 (D) 5

#### (Matching List Sets)

This section contains **FOUR (04)** Matching List Sets. Each set has **ONE** Multiple Choice Question. Each set has **TWO** lists: List-I and List-II. List-I has Four entries (P), (Q), (R) and (S) and List-II has Five entries (1), (2), (3), (4) and (5). FOUR options are given in each Multiple Choice Question based on List-I and List-II and ONLY ONE of these four options satisfies the condition asked in the Multiple Choice Question.

- Match the Column List – I List – II Equation of tangent of conic is given Equation of tangent is given (P)  $y^{2}_{-}=1$ (1)  $x + y = \sqrt{7}$ 9 25 (2) (Q)  $x + y = 3\sqrt{2}$  $(y-3)^{2}$ (x – 1) = 1  $5x + 3y = 15\sqrt{2}$ (R) (3) $(x + y - 1)^2$  $(x-y+3)^2$ 18 32 (S) (4)  $x + y = 4 + \sqrt{5}$ x<sup>2</sup> · = 1 3  $x + y = 4 - \sqrt{3}$ (5) (B)  $P \rightarrow (3); Q \rightarrow (4); R \rightarrow (1); S \rightarrow (2)$ (A)  $P \rightarrow (3); Q \rightarrow (4); R \rightarrow (2); S \rightarrow (1)$ (C)  $P \rightarrow (3); Q \rightarrow (1); R \rightarrow (4); S \rightarrow (2)$ (D)  $P \rightarrow (3); Q \rightarrow (1); R \rightarrow (4); S \rightarrow (1)$
- 9. Match the Column

		List – I	List – II		
	(P)	Minimum value of $\cos^2 \theta + \sec^2 \theta$ is	(1)	2	
	(Q)	Minimum value of $\frac{a+b}{c} + \frac{b+c}{a} + \frac{c+a}{b}$ is	(2)	20	
		(where a,b,c ∈ R')			
	(R)	If $a_1, a_2, a_3, \dots, a_{10} \in \mathbb{R}^+$ such that $a_1.a_2.a_3, \dots, a_{10} = 2^9$ then	(3)	1	
		minimum value of $a_1 + a_2 + a_3 + \dots + 2a_{10}$ is			
	(S)	If $x, y \in \mathbb{R}^+$ such that $2x + 3y = 5$ then maximum value of	(4)	10	
		x <sup>2</sup> y <sup>5</sup> is			
Ī			(5)	6	
	(A)	$P \rightarrow (1); Q \rightarrow (2); R \rightarrow (2); S \rightarrow (1)$ (B) $P \rightarrow (1); Q \rightarrow (5)$	; R→(2	2); S→(3)	
	(C)	$P \rightarrow (3); Q \rightarrow (1); R \rightarrow (4); S \rightarrow (2) \qquad (D) P \rightarrow (3); Q \rightarrow (4)$	; R→(2	2); S→(1)	

Space For Rough Work

8.

10. Equation of hyperbola is xy - 3x + 4y - 16 = 0 then

	List – I	List – II		
(P)	Asymptote equation is	(1)	y - 3 = 0	
(Q)	Transverse axis is	(2)	x + y + 1 = 0	
(R)	Conjugate axis is	(3)	x + 4 = 0	
(S)	Line parallel to directrix	(4)	x - y + 7 = 0	
		(5)	x - y - 7 = 0	

(A) 
$$P \rightarrow (1); Q \rightarrow (2); R \rightarrow (2); S \rightarrow (1)$$
  
(C)  $P \rightarrow (1), (3); Q \rightarrow (4); R \rightarrow (2); S \rightarrow (2)$ 

(B)  $P \rightarrow (3)$ ;  $Q \rightarrow (4)$ ;  $R \rightarrow (1)$ ;  $S \rightarrow (2)$ (D)  $P \rightarrow (3); Q \rightarrow (4); R \rightarrow (2); S \rightarrow (1)$ 

11. Match the Column

List – I								List – II		
(P)	Ordinate	of	foci	of	the	ellipse	(1)	4		
	$4x^2 + 9y^2 +$	⊦8x –18y -	+4=0 is							
(Q)	Abssica of the centre of ellipse $3x^2 + 4y^2 - 6x - 16y = 0$						(2)	2		
	is									
(R)	If e is eccentricity of ellipse $3x^2 + 4y^2 = 1$ then $2e^2 = 1$							1		
(S)	Maximum	distance	between	two	points of	ellipse	(4)	0		
	$5x^2 + 4y^2 =$	=20 is								
							(5)	6		

 $\begin{array}{ll} (A) \quad \mathsf{P} \rightarrow (1); \ \mathsf{Q} \rightarrow (2); \ \mathsf{R} \rightarrow (2); \ \mathsf{S} \rightarrow (1) \\ (C) \quad \mathsf{P} \rightarrow (3); \ \mathsf{Q} \rightarrow (1); \ \mathsf{R} \rightarrow (4); \ \mathsf{S} \rightarrow (2) \end{array} \begin{array}{ll} (B) \quad \mathsf{P} \rightarrow (3); \ \mathsf{Q} \rightarrow (4); \ \mathsf{R} \rightarrow (1); \ \mathsf{S} \rightarrow (2) \\ (D) \quad \mathsf{P} \rightarrow (3); \ \mathsf{Q} \rightarrow (3); \ \mathsf{R} \rightarrow (3); \ \mathsf{S} \rightarrow (1) \end{array}$ 

### (PART – <mark>B)</mark>

### (Non – Negative Integer)

- The sum of first 24 terms of an A.P a1,a2,a3,....; if it is known that 1.  $a_1 + a_5 + a_{10} + a_{15} + a_{20} + a_{24} = 225$ , is equal to
- Three non-zero numbers a, b and c are in A.P.. Increasing a by 1 or increasing c by 2 the 2. number become in G.P., then  $\frac{b}{3}$  equals to
- If the tangents on the ellipse  $4x^2 + y^2 = 8$  at the points (1, 2) and (a, b) are perpendicular to 3. each other, then 17 a<sup>2</sup> is equal to:
- If e and e' be the eccentricities of a hyperbola and its conjugate then the value of  $\frac{1}{e^2} + \frac{1}{e^{2}} =$ 4.
- 5. The product of the lengths of perpendiculars drawn from focus to any tangent to the hyperbola  $9x^2 - 16y^2 = 144$  will be
- If a,b,c  $\in \mathbb{R}^+$  such that abc = 1 then minimum value of (a+b)(b+c)(c+a) is 6.

<b>FIITJEE INTERNAL TEST</b>										
BA	TCHES	– PANI	NI42	6-G1. A	1. A2	& PANI	NI426	6-B1		
		C	Com	mon Tes	st – 5					
Code: 100856										
				ISWED KE						
ANSWER KETS Physics										
				PART – A						
1.	D	2.	В	3.	С	4.	D			
5.	BCD	6.	AB	7.	CD	8.	В			
9.	D	10.	A	11.	В					
1	2	2	16	PARI – B	6	1	1			
5.	2 1	2. 6.	3	Э.	U	4.	4			
0.	·	0.	Ŭ							
				Chemistry	/					
1	D	2	D	PART – A	٨	4	Б			
۱. ح	Б	2. 6		3. 7	BCD	4. 8	Δ			
9.	В	10.	A	11.	D	0.	7.			
				PART – B						
1.	8	2.	20	3.	2	4.	4			
5.	1	6.	25							
PART – A										
1.	В	2.	С	3.	В	4.	С			
5.	С	6.	AB	7.	CD	8.	А			
9.	В	10.	С	11.	D					
	PART – B									
1.	900	2.	4	3.	2	4.	1			
5.	9	6.	Ø							